Ecotoxicology TP Course

Concepts, Tests & Biomarkers

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TP8 Effect and defense biomarkers

PHOTOBIOLOGY



Control



10 ug/L Glifosato



250 ug/L Glifosato



500 ug/L Glifosato

PHASE I: BIOTRANSFORMATION

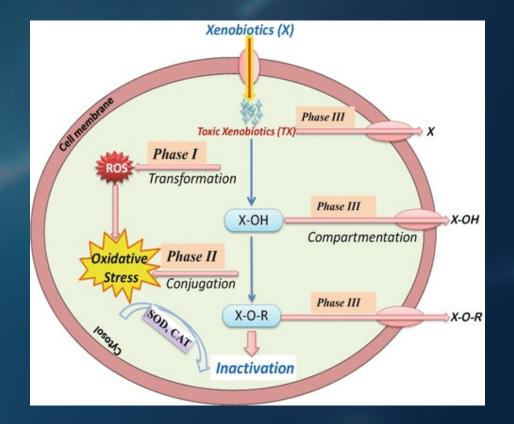
Phase I enzymes and mechanisms **transform xenobiotics into less harmful molecules**, but that may have a ROS-generating potential (for e.g.: CYPA1 enzymes).

PHASE II: CONJUGATION

Phase II mechanisms are composed by enzymatic and non-enzymatic **anti-oxidant mechanisms** that work in conjugation to quench the ROS generated directly by the xenobiotic or by the Phase I biotransformed xenobiotic.

PHASE III: EXCRETION

Phase III mechanisms are based in **membrane proteins that can excrete** directly or throughout vesicle compartments the transformed and/or inactivated xenobiotic to the extracellular environment.



Pyrogaloll + 20_2

Time (seconds)

PRINCIPLES OF THE METHOD:

- Pyrogaloll (Pyr) has a absorption peak at 325 nm.
- Pyrogaloll has an auto-oxidation (degradation) rate.
- The method measures the inhibition of oxidized Pyrogaloll for 120 seconds at 325 nm by Superoxide Dismutase.



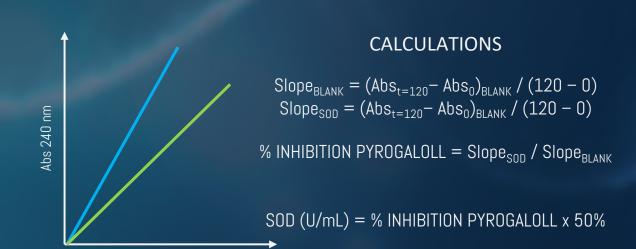
- 1. 100 μg protein (calculate the extract volume needed).
- Remaining volume with PBS buffer.

To zero the spectrophotomer.



SOD ASSAY

- 1. 100 μg protein (calculate the extract volume needed).
- 2. Remaining volume with PBS buffer.
- 3. 0.3 mM Pyrogaloll



 $0_2 + H_2 0_2$

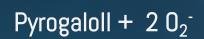
Pyrogaloll +
$$20_2^-$$



$$0_2 + H_2 0_2$$

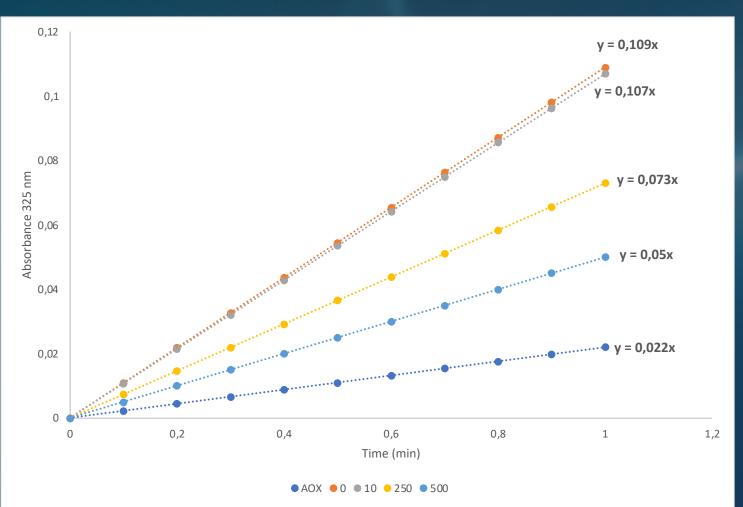
- 1. Extract the proteins with Extraction PBS Buffer (1 mL)
- 2. Centrifuge to precipitate cellular debris
- 3. Use the extract for spectrophotometric assay (60 sec at 325 nm)

	Blank	Reaction
Ultra-pure Water	360 μL	360 μL
Buffer	630 μL	550 μL
Extract	10 μL	10 μL
Pyrogalol	0 μL	80 μL



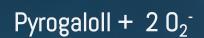


 $0_2 + H_2 0_2$



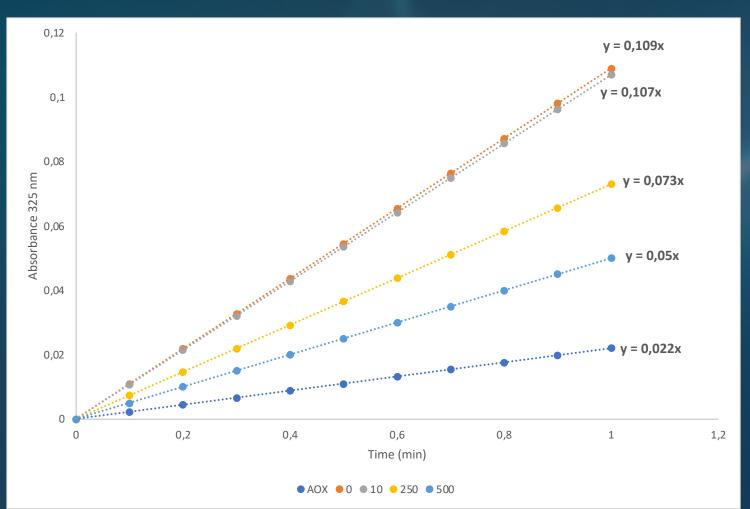
SOD Activity = Enzymatic Rate / (0.5 x AOX Rate)

[Glifosate]	Rate	Activity
0	0.109	
10	0.107	
250	0.073	
500	0.050	
AOX	0.022	



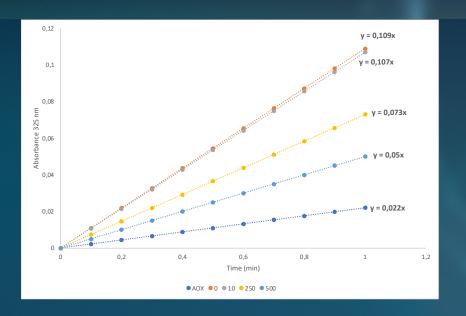


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SOD Activity = Enzymatic Rate / (0.5 x AOX Rate)

[Glifosate]	Rate	Activity
0	0.109	9.91
10	0.107	9.73
250	0.073	6.64
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AOX	0.022	

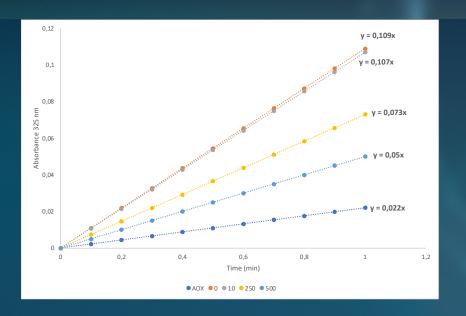


Pyrogaloll +
$$2 O_2^-$$
 SOD $O_2 + H_2O_2$

SOD Activity = Enzymatic Rate /
$$(0.5 \times AOX \text{ Rate})$$

Reaction volume = 10 uL

[Glifosate]	Rate	Activity	[Protein] ug/mL	SOD (U ug Prot)
0	0.109	9.91	208	
10	0.107	9.73	220	
250	0.073	6.64	270	
500	0.050	4.55	186	
AOX	0.022			

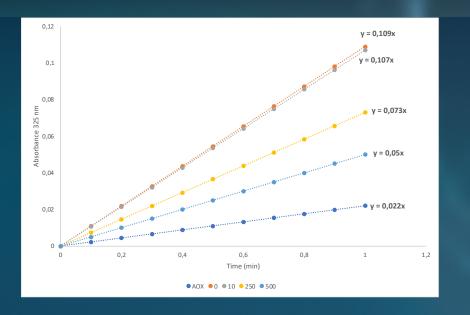


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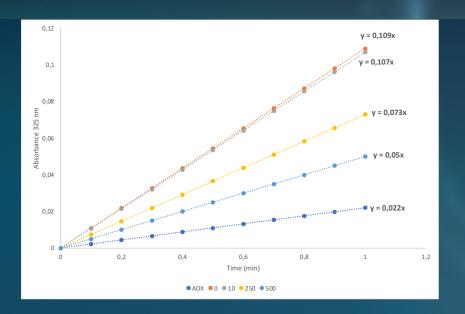
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10	0.107	9.73	220	
250	0.073	6.64	270	
500	0.050	4.55	186	
AOX	0.022			



Pyrogaloll +
$$2 O_2^ O_2 + H_2O_2$$

SOD Activity = Enzymatic Rate / $(0.5 \times AOX \text{ Rate})$ Reaction volume = 10 uL

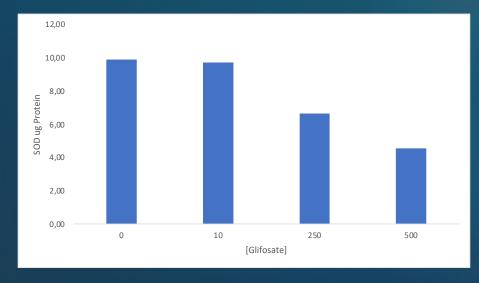
[Glifosate]	Rate	Activity	[Protein] ug/mL	SOD (U ug Prot)
0	0.109	9.91	208	4.76
10	0.107	9.73	220	4.42
250	0.073	6.64	270	3.90
500	0.050	4.55	186	2.44
AOX	0.022			



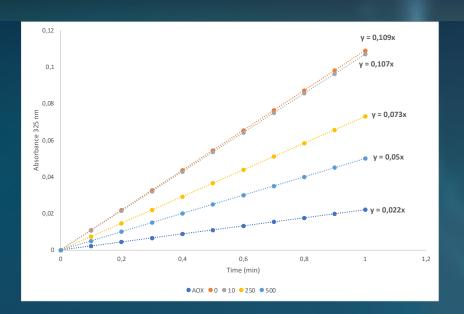
Pyrogaloll +
$$2 O_2^ O_2 + H_2O_2$$

SOD Activity = Enzymatic Rate / (0.5 x AOX Rate)

Reaction volume = 10 uL



[Glifosate]	SOD (U ug Prot)	% Inhibition
0	4.76	
10	4.42	
250	3.90	
500	2.44	



Pyrogaloll +
$$20_2$$

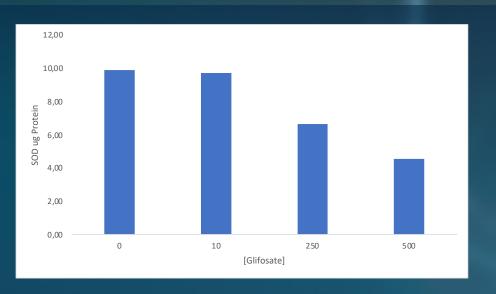


$$0_2 + H_2 0_2$$

SOD Activity = Enzymatic Rate / $(0.5 \times AOX \text{ Rate})$ Reaction volume = 10 uL

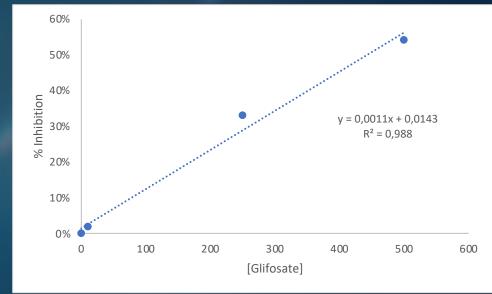
12,00					
10,00					
00,8 6in					
SOD ug Protein					
OS 4,00					
2,00					
0,00					
	0	1	0 [Glifosate]	250	500

[Glifosate]	SOD (U ug Prot)	% Inhibition
0	4.76	0
10	4.42	2
250	3.90	33
500	2.44	54



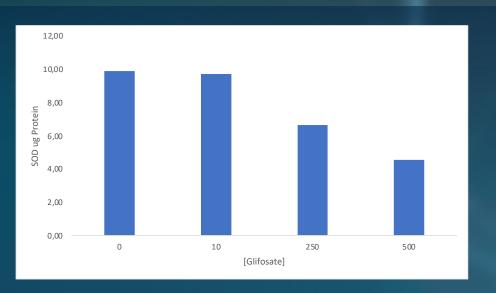
Pyr





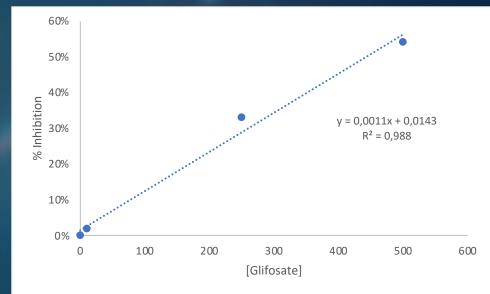
$$IC_{50} = ?$$

[Glifosate]	SOD (U ug Prot)	% Inhibition
0	4.76	0
10	4.42	2
250	3.90	33
500	2.44	54



Pyr
,

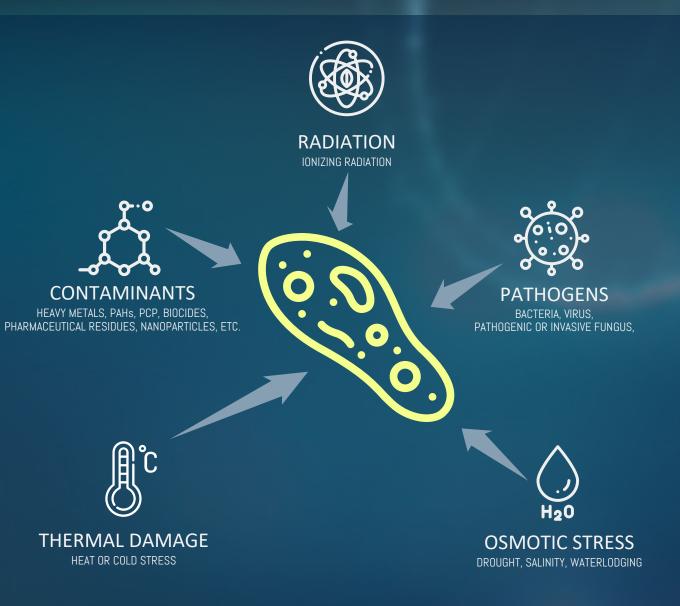




$$IC_{50} = 44.15 \text{ ug/L}$$

[Glifosate]	SOD (U ug Prot)	% Inhibition
0	4.76	0
10	4.42	2
250	3.90	33
500	2.44	54

SOURCES OF OXIDATIVE STRESS





OXIDATIVE STRESS

OXIDATIVE STRESS:

- Oxidative stress reflects an **imbalance** between the systemic **manifestation of reactive oxygen species** and a **biological system's ability to**readily detoxify the reactive intermediates or to repair the resulting damage.
- <u>Disturbances in the normal redox state of cells can cause toxic effects through the production of peroxides and free radicals that damage all components of the cell, including proteins, lipids, and DNA.</u>
- Oxidative stress from oxidative metabolism causes base damage, as well as strand breaks in DNA. Base damage is mostly indirect and caused by reactive oxygen species (ROS) generated, e.g. 0^{2-} (superoxide radical), OH (hydroxyl radical) and H_2O_2 (hydrogen peroxide).
- Further, some reactive oxidative species act as cellular messengers in redox signalling. Thus, oxidative stress can cause disruptions in normal mechanisms of cellular signalling.
- IMPORTANT: ROS are generated during the normal cell functioning thus the cells also developed anti-oxidant mechanisms to counteract the negative effects produced by ROS.
- In Ecotoxicology, the production or activity of these ROS counteractive measures as well as the ROS-biomolecules reaction products are used to evaluate the cell oxidative stress level.

ENZYMES ر م ن ن Typically oxidative stress triggered enzymes, which the activity is related to the stress level imposed. **DEGRADATION PRODUCTS** Molecules or compounds that result from the destruction of a cellular structure or metabolite in reaction to stress. NON-ENZYMATIC ANTI-OXIDANT **MOLECULES** Molecules or compounds that are produced

in reaction to oxidative stress.

BIOMARKER

A trait or molecular entity that can be

measured experimentally and indicate the occurrence of a certain function

(normal or pathological) of a certain

organisms towards a specific stressor.

Ideally these biomarkers should produce a dose related response towards the stressor applied.

ANALYTICAL PROTOCOL

The DNA strand breaks are measured using the DNA alkaline precipitation assay as described by Olive, 1988. The method includes the precipitation of SDS associated nucleoproteins and genomic DNA followed by the determination of the DNA concentration that remains in the supernatant, which constitutes the damaged DNA. This damaged DNA is labelled with a fluorescent dye and the fluorescence is measured using 360:460 nm to excitation:emission wavelength filters.

SAMPLE PREPARATION

- Extract the sample with 0.5 mL Phosphate Buffer and ultrasounds.
- Centrifuge the sample (fast spin).
- Add 50 uL extract (or 50 uL PBS for the blanks) + 250 uL Reaction Mixture + KCl
- Heat the sample for 10 min at 60 °C.
- Place on ice for 10 min.
- Centrifuge at 8000 x g for 5 min.

ANALYTICAL PROTOCOL

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SAMPLE ANALYSIS

- Add 50 uL of the reaction extract to the well.
- Add 200 uL of Hoesch dye
- Read the fluorescence with 360 nm Excitation and 460 nm Emission
- Compare with the DNA standard curve